MINOR ALKALOID FROM COSCINIUM FENESTRATUM*

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Key Word Index—Coscinium fenestratum; Menispermeaceae; stem; 12,13-dihydro-8-oxo-berberine; 13C NMR.

Abstract—Coscinium fenestratum furnished a new minor alkaloid, 12,13-dihydro-8-oxo-berberine, [5,6,13,13a-tetrahydro-9,10-dimethoxydibenzo(a,g)1,3-benzodioxolo(5,6a)quinalizine-8-one], as well as berberine, oxyberberine, tetrahydroberberine (canadine), sitosterol and stigmasterol.

INTRODUCTION

Coscinium fenestratum (N. O. Menispermeaceae) is well known for its medicinal properties [1-4]. Recent studies in our laboratory showed that it possesses marked antihypertensive activity. Recently its activity against tetanus has also been investigated [5].

RESULTS AND DISCUSSION

A literature review indicated the presence of several alkaloids in sister species of *C. fenestratum* [6, 7]. Our studies established that the stem of the plant is rich in berberine (5%). Apart from 12,13-dihydro-8-oxo-berberine five other compounds have also been isolated and identified as berberine, oxyberberine, tetrahydroberberine (canadine), sitosterol and stigmasterol.

Compound 1, mp 218°, analysed for $C_{20}H_{19}N$ O₅ ([M]⁺m/z 353) and has been characterized as 5,6,13,13a-tetrahydro-9,10-dimethoxy dibenzo(a,g)1,3-benzodioxolo(5,6a)quinalizine-8-one (12,13-dihydro-8-oxo-berberine) from the following studies.

In its ¹H NMR characteristic protoberberine signals for methoxyl, methylene dioxy and four aromatic protons were observed. The signals for the H-6a proton appeared at $\delta 3.79$ as a singlet and H-6e as doublet at $\delta 4.95$. The benzylic protons appeared at $\delta 2.82$ and 2.98, respectively. The position of H-13a was located at $\delta 4.61-4.78$ as a double doublet. The relationship between H-13a and the benzylic proton was established by double irradiation experiments, where on irradiating the latter signal a singlet for H-13a and H-6a was produced. Similarly irradiation of signals for H-6a and H-6e resulted in the simplification of the benzylic proton signals.

When 1 was treated with alkaline potassium permanganate it formed noroxyhydrostane (3) (identified by lit. comparison [8,9]) thus establishing the presence of a methylenedioxy group in ring A. The structure of 1 was further confirmed by its partial transformation from tetrahydroberberine [10,11]. Controlled oxidation of tetrahydroberberine (2) with potassium permanganate in acetone resulted in the formation of several oxidation products. Purification by CC gave 1 as one of the minor

$$1 R^1, R^2 = 0$$

$$2 R^1, R^2 = H$$

Table 1. ¹³C NMR spectra of compounds 1 and 2

Chemical shift (CDCl ₃)		
C	1	2
	4040 1	4040 1
1	106.0 d	104.0 d
2	147.0 s	145.2 s
3	147.0 s	145.4 s
4	108.5 d	107.0 d
4a	129.0 s	127.7 s
5	39.1 t	35.0 t
6	39.1 t	51.0 t
8	$163.0 \ s$	53.4 t
8a	122.8 s	126.2 s
9	$152.2 \ s$	144.1 s·
10	149.5 s	149.4 s
11	115.5 d	110.0 d
12	121.4 d	123.0 d
12a	131.0 s	126.2 s
13	38.0 t	28.1 t
13a	61.2 d	55.2 d
13b	129.0 s	129.0 s
-0		
-OCH ₂	101.0 t	100.9 t
-ОМе	55.0 q	59.0 q
-ОМе	56.1 q	59.2 q

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products (5%) together with noroxyhydrostane (3) as the major oxidation product. ¹³C NMR signal assignments for structure (1) were also made by comparison with data for tetrahydroberberine (2) (Table 1). Recently ¹³C NMR data for berberine and sanguinarine has been recorded by C. Blaskó et al. [12]. The mass spectral fragmentation of the compound also supported the proposed structure.

Recently a compound namely lemiocranine [13] with the same molecular formula and related features has been isolated from *B. darwinii*. However, comparison of the spectral data of 1 and lemiocranine, showed that they possess different structures.

EXPERIMENTAL

General. Mps: uncorr. ¹H NMR were recorded at 60 and 90 MHz.

Plant material. Stems of C. fenestratum Colebr. were collected from the Attapadi forest in Western Ghats, Kerala, India and identified by AICRPE Division, RRL, Jammu. The CHCl₃ extract of the dried and powdered plant material was subjected to CC on silica gel G.

Compound (1). Mp 218 °C, $[\alpha]_{D}^{28}$ (MeOH) – 40, analysed for $C_{20}H_{19}N$ O₅ (observed C 67.8, H 5.2, N 3.8, required C 67.98, H 5.38 and N 3.96%). $\lambda_{\max}^{\text{MeOH}}$ 277 nm, –ve Cotton effect; ν_{\max}^{KBF} cm⁻¹, 2945, 1645, (C=O), 1540, 1500, 1388, 1240, 1160

(OMe), 1020, 930,
$$\begin{pmatrix} -O \\ -O \end{pmatrix}$$
 CH₂ and 910. ¹H NMR (CDCl₃)

δ: 6.95 (s, 2H, H-11 and H-12), 6.70 (s, 2H, H-1 and H-4), 5.90

$$\left(s, 2H, \frac{-O}{-O}\right)$$
 CH₂ 4.95 (d, 1H, J = 5 Hz, H-6e), 4.61-4.78 (dd,

1H, J = 5 and 11 Hz, H-13a), 4.0 (s, 3H, -OMe), 3.84 (s, 3H, -OMe), 3.79 (s, 1H, H-6a), 2.98 (d, 2H, J = 5 Hz, H-13), 2.82 (s, 2H, H-5). MS m/z 353 [M]⁺ 338, 310, 296, 281, 267, 178 (66%), 176, 148, 144.

Alkaline oxidation. To a soln of 1 (25 mg) in dry Me₂CO (3 ml) and 10% NaOH (1 ml), KMnO₄ (40 mg) in Me₂CO was added with continuous stirring at room temp. After one hr of stirring, the reaction mixt. was worked-up to afford compound 3, identified as nor-oxyhydrostane, mp 188° (lit. 188) [M]⁺ m/z 191, ¹H NMR CCl₄ δ : 7.4 (br s, 1H, H-1), 6.6 (br s, 1H, H-4),

6.0 (s, 2H,
$$\stackrel{\text{O}}{-\text{O}}$$
CH₂), 3.98 (br s, 2H, H-6), 3.0-2.9 (br s, 2H,

Partial oxidation of tetrahydroberberine. THB (250 mg), in Me₂CO (50 ml) was refluxed with KMnO₄ for 45 min and then worked-up as usual. Sepn by CC afforded different products and one of them was identified as 1 by direct comparison with the isolated natural compound (co-TLC, mmp and spectral studies); yield 5%.

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